Seasonal variability of atmospheric nitrogen oxides and non-methane hydrocarbons at the GEOSummit station, Greenland

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Abstract. Measurements of atmospheric nitrogen oxides NOx (NOx = NO + NO2), peroxyacetyl nitrate (PAN), NOy, and non-methane hydrocarbons (NMHC) were taken at the Greenland Environmental Observatory at Summit (GEOSummit) station, Greenland (72.34° N, 38.29° W; 3212 m a.s.l.), from July 2008 to July 2010. The data set represents the first year-round concurrent record of these compounds sampled at a high latitude Arctic site. Here, the study focused on the seasonal variability of these important ozone (O3) precursors in the Arctic troposphere and the impact from transported anthropogenic and biomass burning emissions. Our analysis shows that PAN is the dominant NOy species in all seasons at Summit, varying from 42 to 76%; however, we find that odd NOy species (odd NOy = NOy − PAN − NOx) contribute a large amount to the total NOy speciation. We hypothesize that the source of this odd NOy is most likely alkyl nitrates and nitric acid (HNO3) from transported pollution, and photochemically produced species such as nitrous acid (HONO).

FLEXPART retroplume analyses and black carbon (BC) tracers for anthropogenic and biomass burning (BB) emissions were used to identify periods when the site was impacted by polluted air masses. Europe contributed the largest source of anthropogenic emissions during the winter months (November–March) with 56 % of the total anthropogenic BC tracer originating from Europe in 2008–2009 and 69 % in 2009–2010. The polluted plumes resulted in mean enhancements above background levels up to 334, 295, 88, and 1119 pmol mol−1 for NOy, PAN, NOx, and ethane, respectively, over the two winters. Enhancements in O3 precursors during the second winter were typically higher, which may be attributed to the increase in European polluted air masses transported to Summit in 2009–2010 compared to 2008–2009. O3 levels were highly variable within the sampled anthropogenic plumes with mean O3 levels ranging from −6.7 to 7.6 nmol mol−1 during the winter periods.

North America was the primary source of biomass burning emissions during the summer; however, only 13 BB events were observed as the number of air masses transported to Summit, with significant BB emissions, was low in general during the measurement period. The BB plumes were typically very aged, with median transport times to the site from the source region of 14 days. The analyses of O3 and precursor levels during the BB events indicate that some of the plumes sampled impacted the atmospheric chemistry at Summit, with enhancements observed in all measured species.

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1 Introduction

The seasonality of ozone (O$_3$) and its precursors for photochemical production, such as nitrogen oxides, i.e. NO$_x$ (NO$_x$ = NO + NO$_2$), peroxyacetyl nitrate (PAN), and non-methane hydrocarbons (NMHC), in the remote Arctic troposphere is governed by a combination of transport pathways, photochemistry, and stratospheric influx (Klonecki et al., 2003; Stohl et al., 2006; Law and Stohl, 2007; Liang et al., 2011). Improving our knowledge on the seasonality of O$_3$ and its precursors and the relative importance of source regions and transport variability is essential as recent studies have suggested that tropospheric O$_3$ may have a large impact on radiative forcing and climate feedbacks in the Arctic region (Shindell et al., 2006; Shindell, 2007; Quinn et al., 2008).

Polluted air masses originating from anthropogenic and biomass burning sources in the mid-latitude regions can transport long-lived reservoir species of NO$_x$, such as PAN, and nitric acid (HNO$_3$), to the Arctic region (Wofsy et al., 1992; Wespes et al., 2012), which may reform NO$_x$ and result in enhanced levels far downwind from the emission sources (Beine et al., 1997; Walker et al., 2010). NMHC may also be transported in air masses from anthropogenic and biomass burning sources. The mole fractions of NMHC in the Arctic atmosphere can vary greatly during the year due to seasonal variability in emissions, transport pathway variability, and the reaction with OH radicals (Jobson et al., 1994; Blake et al., 2003; Swanson et al., 2003).

Studies of pollution plumes with airborne, satellite- and ground-based observations, and model simulations show that long-range transport from Europe and North America to the lower Arctic troposphere may constitute a large source of tropospheric O$_3$ and O$_3$ precursors, whereas at higher altitude, pollution plumes transported from Asia are an important source (e.g. Atlas et al., 2003; Klonecki et al., 2003; Lamarque and Hess, 2003; Law and Stohl, 2007; Shindell et al., 2008; Fisher et al., 2010; Singh et al., 2010; Walker et al., 2012; Wespes et al., 2012; Bian et al., 2013). A large contribution to the seasonality of O$_3$ and O$_3$ precursors in the Arctic troposphere is due to variability in the location of the Arctic polar front (Klonecki et al., 2003; Stohl, 2006). During winter in the Northern Hemisphere, the polar front expans southward over North America, Europe, and Siberia allowing for direct transport of polluted air masses from sources within these latitudes to the Arctic. The Arctic polar front recedes in summer, reducing the impact of these pollution sources on the Arctic lower troposphere. However, it has been shown that the transport of emissions from biomass burning regions to the Arctic is possible during summer (Stohl, 2006) and that they can strongly impact the atmosphere above GEOSummit station (Stohl et al., 2006). Results from a modelling study by Walker et al. (2012) using tagged emissions in the global chemical transport model GEOS-Chem show that during summer the primary emissions that impact the production of O$_3$ in the Arctic region were from high latitude regions, whereas, during the fall and winter periods, transport of emissions from mid-latitude regions in North America and Europe is possible.

A number of studies have discussed the seasonality of surface O$_3$ (Bottenheim et al., 1994; Beine et al., 1997; Monks, 2000; Browell et al., 2003; Helmig et al., 2007b; Walker et al., 2012), nitrogen oxides (Barrie and Bottenheim, 1991; Honrath and Jaffe, 1992; Bottenheim et al., 1994; Muthuramu et al., 1994; Beine et al., 1997; Solberg et al., 1997; Dibb et al., 1998; Munger et al., 1999; Beine and Krognes, 2000; Stroud et al., 2003; Thomas et al., 2011), and NMHC (Jobson et al., 1994; Blake et al., 2003; Klonecki et al., 2003; Swanson et al., 2003; Helmig et al., 2014) in the Arctic. However, due to the logistical difficulties in measuring at remote Arctic locations, the majority of seasonal studies have taken place at coastal sites in northern Europe, Canada, and Alaska, or focused on the late spring/summer periods. Seasonal and interannual studies of nitrogen oxides in the remote Arctic troposphere are largely missing. The high latitude Arctic has negligible impact from local pollution sources, and local production of NO$_x$ from PAN decomposition is expected to be small in this cold region. Therefore, enhanced mole fractions of nitrogen oxides are primarily a result of long-range transported pollution from anthropogenic or biomass burning sources in Europe, North America, and Asia, or of downward transport from the stratosphere (e.g. Liang et al., 2011). Measurements within the boundary layer over Greenland ice sheet are also influenced by emissions from photochemical reactions within the snowpack (Garnas et al., 2007, and references therein), and variability in the boundary layer height (Cohen et al., 2007; Anderson and Neff, 2008; Van Dam et al., 2013), both of which may impact the observed NO$_x$ budget and seasonal cycle of O$_3$ precursors in this region.

A build-up of O$_3$ precursors during winter in the Arctic free troposphere may have important implications for the tropospheric O$_3$ budget in the mid-latitudes during late spring and early summer (Gilman et al., 2010). Modelling studies have postulated that air masses originating from the Arctic region can result in the transport of NO$_x$ and NMHC to the North Atlantic and enhance tropospheric O$_3$ in this region due to the thermal decomposition of PAN (Honrath et al., 1996; Hamlin and Honrath, 2002).

This study utilizes 2 years of continuous measurements and model results to characterize the seasonally varying magnitude of O$_3$ and its precursors in the remote high latitude Arctic and potential impact from transported pollution. Year-round measurements of NO$_x$, NO$_y$, PAN, O$_3$, and NMHC from the high altitude Greenland Environmental Observatory at Summit (GEOSummit) station in Greenland are presented. The paper is structured as follows: in Sect. 2, the techniques to measure NO, NO$_2$, NO$_y$ (total reactive nitrogen oxides NO$_y$ = NO + NO$_2$ + PAN + HNO$_3$ + HONO + others), PAN, and NMHC are presented and the FLEXPART Lagrangian

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particle dispersion model utilized in this study is discussed. Sect. 3.1 presents the seasonal cycles of O₃ precursors at the measurements site and the NO₃ speciation is investigated. In Sect. 3.2, the source contributions to enhanced O₃ precursors from anthropogenic emissions and biomass burning are discussed. Finally, a summary of the main findings is given in Sect. 4.

2 Experimental methods

2.1 GEOSummit station

Measurements of NOₓ, NO₃, PAN, and NMHC were performed at the GEOSummit station (hereafter called Summit), Greenland (72.34º N, 38.29º W; 3212 m a.s.l.), from July 2008 to July 2010. Inlets for the instruments were installed ~8.5 m above the snowpack (~6.5 m toward the end of the measurement period, as a result of snow accumulation) on a meteorological tower located approximately 660 m south-west of the main camp within the “clean-air” sector. Tubing and cables were routed through a heated pipe to a buried laboratory facility.

2.2 Measurements

2.2.1 Nitrogen oxides

Measurements of NO, NO₂, and NO₃ were performed with an automated O₃ chemiluminescence detection system (Ridley and Grahek, 1990). The instrument was developed at Michigan Technological University and is based on the same design that was used in Newfoundland in 1996 (Peterson and Honrath, 1999), subsequently installed at Summit during campaigns in 1998, 1999, and 2000 (Honrath et al., 1999, 2002; Dibb et al., 2002), and at the Pico Mountain Observatory from 2002 to 2010 (Val Martin et al., 2006). NO₂ and NO₃ were detected by chemiluminescence after reduction to NO using a photolytic NO₂ converter (Kley and McFarland, 1980) and a gold-catalyzed NO₃ converter in the presence of CO (Bollinger et al., 1983; Fahey et al., 1985), respectively. NO₃ is given as the sum of reactive nitrogen oxides. In the Arctic, NO₃ is primarily comprised of NO, NO₂, PAN, HNO₃, nitrous acid (HONO), and particulate nitrate (p-NO₃). For the instrument used in this study, a photolytic blue LED NO₂ converter (Air-Quality Design Inc., Colorado) was installed. Photolytic converters have lower conversion efficiencies than molybdenum converters; however, interferences from other species photolyzing to NO, such as HONO and PAN, are reduced (Pollack et al., 2011; Villena et al., 2012). The sample mass flow controllers (MFC) and the NO₂ and NO₃ converters were housed inside the inlet box on the tower to minimize the residence time of NO₃ species inside the PFA (perfluoralkoxy) tubing.

During each measurement cycle of 10 min, the NO and NO₂ signals were recorded as 30 s averages and NO₃ signals as 20 s averages, after a period of equilibration in each mode. Zero measurements of NO were performed at the start and end of each measurement cycle by mixing O₃ with the sample upstream of the reaction chamber. The zero signals were measured to determine the interference signal in the reaction chamber, which was then subtracted from the measured signals. Calibrations were performed every 12 h to determine the sensitivity of the instrument to NO via standard addition (10 cm³ min⁻¹) of NO in nitrogen (N₂) (ranging from 0.792 to 0.930 mmol mol⁻¹, Scott Marrin, Scott Specialty Gases) to the sample flow of 650 cm³ min⁻¹ at the inlet on the tower. A known amount of NO₂, generated via gas phase titration of NO with O₃, was also added to the sample flow during the calibration cycle to determine the conversion efficiencies of the NO₂ and NO₃ converters. In addition to the standard calibrations, every 3 days the conversion efficiency of the NO₂ converter to HNO₃ and n-propyl nitrate (PNP) were determined and artifacts for NO₃, NO, and NO₂ were measured via sampling NO₂-free air (breathing air grade, Airgas, Radnor, PA, USA). The final data sets were corrected for this artifact.

Approximately 7% of the final NO and NO₂ data sets and 8% of the final NO₃ data set were removed due to known instrument issues. Additional filtering procedures were applied to remove points potentially contaminated by local pollution. Variability in the 20 and 30 s averaged data was compared to the expected value from photon counting statistics which are treated as a Poisson distribution. Measurements with variability greater than 3 times the Poisson value were then removed from the final data set (~5%). Evaluation of these periods show that they typically occur when the wind direction was from the main camp, confirming that local pollution was the main source of the variability. Erroneous data points as a result of unknown instrument issues, or periods when the skiway was groomed, were also removed. Less than 0.2% of the total data were classified as erroneous and each point was manually checked by comparing to adjacent observations. Finally, large negative mole fractions, as a result of large variability between modes during the measurement cycle, were removed (~1%).

The final NO, NO₂, NO₃, and NO₄ data used in this work were further averaged over a 30 min period. NO₃ was determined as the sum of the NO and NO₂ measurements during each 10 min cycle. The overall uncertainty for the 30 min data is calculated from the root sum of the squares of the measurement accuracy, artifact uncertainty, and precision. Maximum uncertainties for NO, NO₂, and NO₃ at 50 pmol mol⁻¹ are 10, 17, and 19%, respectively. For NO₃, the total uncertainty is also dependent on the conversion efficiencies of the NO₃ species, which is estimated to be ~10% based on the conversion efficiencies and NO₃ levels expected at Summit. The total uncertainty in NO₃ is estimated to be ~12% at 200 pmol mol⁻¹. Detection limits for the 30 min averages were estimated from the 2σ precision of the mea-
measurements, based on the standard error of the photon counting noise. Median detection limits for NO, NO₂, NO₃, and NOₓ were 2.7, 5.0, 5.7, and 3.8 pmol mol⁻¹, respectively. Mole fractions below the detection limit, including small negative mole fractions (as a result of uncertainties in the zero measurement and artifact corrections), were included in all averaging calculations to ensure the final values were not biased. Further details on the calibrations performed and the precision and accuracy of the measurements are provided in the Supplement.

2.2.2 Peroxyacetyl nitrate

A commercial PAN gas chromatography analyzer (PAN-GC, Metcon, Inc., Boulder, CO) was installed alongside the NOₓ仪器 to determine atmospheric PAN mole fractions. The PAN instrument is based on gas chromatography with electron capture detection (GC-ECD). The instrument was equipped with a preconcentration unit to improve the detection limit whilst allowing for PAN sampling every 10 min. The preconcentration unit traps PAN and carbon tetrachloride (CCl₄) on a Peltier-cooled (5 °C) capillary column with subsequent desorption at approximately 35 °C for injection onto the main GC column, which was set to a temperature of 15 °C. Ultra-pure nitrogen gas (99.9999 % purity) was used as the carrier gas for the PAN-GC.

The instrument was calibrated approximately every week using a known concentration of PAN, which was photochemically produced from the same NO-calibration gas used for the NOₓ instrument described in Sect. 2.2.1. The PAN concentration was determined from the NO-calibration gas mixing ratio, the mass flow rates, and the conversion efficiency of NO to PAN, and varied from ~520 to 700 pmol mol⁻¹, during the measurement period. The NO gas was delivered to a reaction cell inside the PAN calibration unit which contained a UV mercury lamp to photolyze an excess of acetone (in zero air) to form peroxyacetyl radicals that oxidize the NO gas to NO₂. NO₂ then reacts with peroxyacetyl radicals to form PAN. The PAN calibration gas was sent to the inlet on the tower and then sampled by the GC-ECD. The NOₓ instrument was used to determine the conversion efficiency of NO to PAN at the beginning and end of the measurement period. Some of the gas sent to the PAN inlet was redirected to the NOₓ instrument and the NO and NO₂ levels were measured with the PAN calibration unit switched on and off. The conversion efficiency remained relatively constant throughout the measurement period at 93 ± 3 % (mean ± uncertainty).

Uncertainty in the PAN calibration is associated with uncertainties in (a) the calculation of the NO addition, (b) the conversion of NO to PAN in the calibration unit, and (c) variability in the PAN peak areas during the calibration. The root mean squared propagation of error gives an uncertainty in the calibration standard of 16 % during normal operation. The sensitivities determined from the weekly PAN calibrations were interpolated to the measurements to take into account any drifting. Additional uncertainty arises from short-term variability in sensitivity between calibrations. CCl₄ was used as an internal reference during periods when calibrations were not taken (Karbiwnyk et al., 2003). The atmospheric concentration of CCl₄ is relatively constant; therefore, any changes in the CCl₄ peak area are primarily caused by changes in the instrument sensitivity. The median relative standard deviation (RSD) of the CCl₄ peak area between calibrations was 6 %. During a period between 14 February 2009 and 25 May 2009 there was a gap in the calibrations caused by a blockage in the tubing that delivered the PAN calibration gas to the inlet. To estimate PAN sensitivities during this period, the calibrations before and after the blockage were used to perform a linear regression between the PAN sensitivities and the CCl₄ peak areas. The slope and intercept from the regression were then used to determine the PAN sensitivities from the CCl₄ peak areas between 14 February 2009 and 25 May 2009. An additional uncertainty in the PAN measurements of 12 % (median RSD) was estimated from errors in the regression. The total uncertainty in the PAN calibration is determined from the root sum of the squares of the uncertainties in the calibrations and the variability, giving an uncertainty of 17 % during normal operation, and 21 % during spring 2009.

The limit of detection (LOD) of the instrument was estimated from the peak to baseline noise ratio. The LOD is defined as the mole fraction giving a signal-to-noise (S/N) ratio of 3. The noise level was determined as the peak-to-peak value of the baseline noise from a region just before the PAN peak, for each chromatogram. The limit of detection was the highest during the first few months of operation with a LOD up to 75 pmol mol⁻¹ in September 2008. The LOD improved after November 2008, decreasing to around 20–40 pmol mol⁻¹ from April 2009 to late January 2010. In spring 2010, the LOD increased to ~40–65 pmol mol⁻¹ as a result of drifting and a noisy baseline as the detector became dirty. Due to the degradation of the detector, no data after 28 April 2010 were included in the analyses here.

Similarly to the NOₓ and NO₂ data, the PAN measurements were averaged over 30 min. The total uncertainty for the 30 min averaged PAN mole fractions was determined from the root sum of the squares of the measurement precision (estimated as 2σN₀.₅, where N is the number of points averaged in 30 min (N ≤ 3)) and from the uncertainty in the calibration. The precision was typically <26 pmol mol⁻¹ with a mean (median) value of 6.5 pmol mol⁻¹ (5.3 pmol mol⁻¹). Using the median precision, the total uncertainty in the PAN measurements at 100 pmol mol⁻¹ was estimated to be approximately 18 % during normal operation, and 22 % during spring 2009. For the statistical analyses presented here, 30 min averages below the LOD were treated as one-half of the limit of detection. Despite the high LOD, ~74 % of the 30 min averaged PAN data were above the LOD during 2008, and >99 % were
above the LOD from 2009 onwards. The final PAN data set was not filtered for wind direction as analyses showed that there was no obvious influence from camp pollution on the PAN measurements.

2.2.3 Non-methane hydrocarbons

NMHC were continuously sampled from June 2008 to July 2010 using a fully automated and remotely controlled GC system that was specifically designed for this study. Details of the set-up at Summit are given in Helmig et al. (2014). The GC is a further development of the instrument operated at the Pico Mountain Observatory and described in detail by Tanner et al. (2006). The NMHC system provided approximately eight ambient measurements of C₂–C₆ hydrocarbons each day, with each measurement representing a collection/sample integration time during the sample pre-focusing step of ∼45 min. In addition, one blank sample was analyzed ∼daily, and a standard approximately every 2 days.

The inlet for the GC instrument was installed on the same tower as the PAN, NOₓ, and NO₃ inlets. The instrument relied on a cryogen-free sample enrichment and injection system. All consumable gases were prepared at the site with a hydrogen generator, compressor, and air purification system. Aliquots of the sample stream were first passed through a Peltier-cooled trap to dry the air to a dew point of −30°C, then through an ozone scrubber, and NMHC were then concentrated on a Peltier-cooled (∼−25°C) multi-stage adsorbent trap. Analysis was accomplished by thermal desorption and injection onto an aluminum oxide (Al₂O₃) porous layer open tubular (PLOT) column for cryogen-free separation on a SRI Model 8610 GC with flame ionization detection (FID). Blanks and standard samples were injected regularly from the manifold. The gravimetric and whole air standards that were used were cross-referenced against our laboratory scale for volatile organic compounds, which has been cross-referenced against national and international scales, including through two previous audits by the World Calibration Centre for Volatile Organic Compounds (WCC-VOC) (http://link-ifu.fzk.de/wcc-voc/). At 100 pptv mole fraction, analytical accuracy and precision were typically better than 3–5%, yielding a combined uncertainty estimate of ∼5%.

The instrument achieved low single digit pmol mol⁻¹ detection limits. During summer, when NMHC levels for C₄–C₅ NMHC at times dropped below the detection limits, for the statistical and whisker plot calculations those data were treated as one-half of the detection limit.

2.2.4 Ozone

Surface O₃ was measured by an O₃ analyzer located in the Temporary Atmospheric Weather Observatory (TAWO) building a few hundred metres from camp by the National Oceanic and Atmospheric Administration (NOAA) as part of the core atmospheric measurements that began in 2000.
Figs. S1 and S2 in the Supplement. The result suggests that although there may be some differences in the magnitude of the FLEXPART tracers, the transport simulated with the BC tracer correlates well with the CO tracer (Pearson’s correlation coefficient, $R = 0.97$); therefore, significant pollution events were unlikely to be missed.

3 Results and discussion

3.1 Seasonal cycles

3.1.1 Reactive nitrogen oxides

Figure 1a-d show the statistical analyses of the monthly averaged NO$_y$, NO$_x$, PAN, and O$_3$ ambient mole fractions, during the measurement period from July 2008 to July 2010. A malfunction with the NO$_y$ instrument resulted in missing NO$_x$ and NO$_y$ data from 24 November 2008 to 30 March 2009. Seasonal cycles are observed for all measured species, with higher levels during the late winter–early spring period and lower mole fractions from summer to fall. The positively skewed whiskers indicate that air masses with elevated levels of NO$_x$, NO$_y$, and PAN were sampled year-round. Anthropogenic and biomass burning emissions transported to the site from North America and Europe are a major source of these enhancements (see Sect. 3.2). Observations from the Arctic Research of the Composition of the Troposphere from Aircraft and Satellites (ARCTAS) mission in 2008 showed that mixed stratospheric–tropospheric air masses, above 5 km, have elevated levels of O$_3$ precursors such as NO$_x$ and HNO$_3$, which can subsequently be converted to PAN (Liang et al., 2011). Therefore, high mole fractions observed in PAN, NO$_x$, and NO$_y$, may also be the result of sampling air masses mixed with those originating from the stratosphere and upper-troposphere.

Table 1 gives a statistical summary for the monthly averages of PAN, NO$_y$, NO$_x$, and O$_3$ during 2008–2010. Maxima in monthly mean mole fractions of PAN occur in April with mean levels of $217 \pm 65$ (1σ) pmol mol$^{-1}$ and $273 \pm 100$ pmol mol$^{-1}$, for 2009 and 2010, respectively. The difference between monthly mean PAN levels observed in April 2010 and April 2009 are within the measurement uncertainties (as shown in Fig. S3, Supplement), however, the 30 min averages (shown in Fig. S4) indicate high levels of PAN, NO$_y$, and NO$_x$ in April 2010. These high levels are explored further in Sect. 3.2.1. Monthly mean NO$_y$ mole fractions also peaked during April 2010 at $352 \pm 102$ pmol mol$^{-1}$. PAN mole fractions at Summit and the magnitude of the PAN springtime peak are consistent with observations at other high latitude sites such as Zeppelin Mountain, Svalbard (Beine et al., 1997; Solberg et al., 1997; Beine and Krognes, 2000), and Alert, Northwest Territories, Canada (Worthy et al., 1994; Dassau et al., 2004). PAN levels in September 2008 were very low and typically below the detection limit as a result of a noisy baseline. In 2009, PAN reached a monthly minimum of 63 pmol mol$^{-1}$ in July. NO$_y$ mole fractions did not decrease as quickly as PAN from spring to summer in 2009 and reached a minimum monthly average during September. We find that the PAN and NO$_y$ summer mole fractions observed here are comparable to previous measurements performed at the same site in 1998 and 1999, when observed PAN levels were typically 20–150 pmol mol$^{-1}$, and NO$_y$ levels ranged between 100 and 300 pmol mol$^{-1}$ (Honrath et al., 1999; Ford et al., 2002). The slower decrease in NO$_y$ from spring to summer, compared to PAN, is a result of the presence of NO$_x$ and odd NO$_y$ (odd NO$_y = NO_y - PAN - NO_x$) over the summer months and is discussed further below.

The seasonal cycle of PAN is governed by the rate of thermal decomposition and transport patterns. The warmer summer temperatures result in the decomposition of PAN during long-range transport; additionally, during the summer months the polar front recedes north, thus reducing the potential for anthropogenic emissions to reach the measurement...
Table 1. Monthly statistics for NO\textsubscript{x}, NO\textsubscript{y}, PAN, O\textsubscript{3}, and the NO\textsubscript{y} budget measured at Summit from 2008 to 2010.

<table>
<thead>
<tr>
<th>Year</th>
<th>Month</th>
<th>NO\textsubscript{x} (pmol mol\textsuperscript{-1})</th>
<th>NO\textsubscript{y} (pmol mol\textsuperscript{-1})</th>
<th>PAN (pmol mol\textsuperscript{-1})</th>
<th>O\textsubscript{3} (pmol mol\textsuperscript{-1})</th>
<th>NO\textsubscript{x} / NO\textsubscript{y}</th>
<th>PAN/NO\textsubscript{x} %</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Mean ± SD</td>
<td>Mean ± SD</td>
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<td>Mean ± SD</td>
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<tr>
<td>2008</td>
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<td>21 ± 18</td>
<td>133 ± 47</td>
<td>59 ± 27</td>
<td>46 ± 5</td>
<td>15 ± 7</td>
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<td>148 ± 66</td>
<td>79 ± 53</td>
<td>45 ± 7</td>
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<td>10 ± 7</td>
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<td>52 ± 31</td>
<td>39 ± 5</td>
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<td>165 ± 35</td>
<td>101 ± 22</td>
<td>40 ± 4</td>
<td>0 ± 5</td>
<td>58 ± 10</td>
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<td></td>
<td>Dec</td>
<td>--</td>
<td>88 ± 27</td>
<td>40 ± 2</td>
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<tr>
<td>2009</td>
<td>Jan</td>
<td>--</td>
<td>84 ± 20</td>
<td>44 ± 3</td>
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<tr>
<td></td>
<td>Feb</td>
<td>--</td>
<td>106 ± 44</td>
<td>48 ± 4</td>
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<td>Mar</td>
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<td>181 ± 77</td>
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<td>217 ± 65</td>
<td>55 ± 6</td>
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<td>May</td>
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<td>56 ± 7</td>
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<td>Jun</td>
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* PAN/NO\textsubscript{x} and NO\textsubscript{x} / NO\textsubscript{y} ratios determined using coincident measurements only.

site (Beine and Krognes, 2000; Stohl, 2006). Measurements have shown that PAN is typically the largest contributor to NO\textsubscript{x} in the Arctic, due to the rapid formation of PAN near the source region and a long lifetime in the free troposphere (Solberg et al., 1997; Munger et al., 1999; Ford et al., 2002; Alvarado et al., 2010; Singh et al., 2010; Liang et al., 2011). However, there have been very few studies on the seasonal variability of the NO\textsubscript{x} speciation in the Arctic due to limited measurements over winter months.

The full annual cycle of NO\textsubscript{x} contributions from PAN and NO\textsubscript{x} during this study provides some information on the NO\textsubscript{x} speciation, year-round, at Summit. Table 1 shows that PAN is the dominant form of NO\textsubscript{x} all year, with monthly mean [PAN]/[NO\textsubscript{x}] ratios above 50 % in spring and fall, reaching a maximum of 76 % in April 2009. The lowest [PAN]/[NO\textsubscript{x}] ratios occurred during the summer, with a minimum monthly mean of 42 % in July 2008. The seasonal cycle for NO\textsubscript{x} does not follow PAN and NO\textsubscript{x} at Summit. As shown in Table 1, monthly mean NO\textsubscript{x} levels peak 1 month later than NO\textsubscript{x} and PAN, coinciding with an increase in solar radiation. Thus, the contribution of NO\textsubscript{x} to NO\textsubscript{y} maximizes over the summer (10–15 %). The monthly average NO\textsubscript{x} contribution decreased to \(<3\ %\) over winter and typically NO\textsubscript{x} levels were below the detection limit of the instrument. The thermal decomposition of PAN is a possible local source of ambient NO\textsubscript{x} during spring and summer months (Beine et al., 1997); however, the ambient temperatures during the measurement period were always below freezing. The PAN lifetime at Summit is nearly 2 days at 0°C (Sander et al., 2013); therefore, the contribution to NO\textsubscript{x} from thermal decomposition at the site is expected to be very low.

Studies have hypothesized that photochemical reactions within the snowpack result in the release of NO\textsubscript{x} and also HONO to the overlying atmosphere (e.g. Honrath et al., 1999, 2000a, b, 2002; Munger et al., 1999; Beine et al., 2002; Dibb et al., 2002; Dominé and Shepson, 2002; Beine et al., 2003; Grannas et al., 2007; Thomas et al., 2011). Thus, the increase of ambient NO\textsubscript{x} with radiation in spring suggests a possible photochemical source. The diurnal cycles of ambient NO\textsubscript{x} at Summit (discussed further below) also indicate a photochemical source of NO\textsubscript{x} from the snowpack.

Figure 2 shows the monthly averaged mole fractions of NO\textsubscript{x} and PAN + NO\textsubscript{x} during the measurement period. The results show that the sum of PAN and NO\textsubscript{x} cannot always
account for the monthly averaged NO$_y$ within the measurement uncertainty range, suggesting a significant source of odd NO$_y$ at Summit. When considering the 30 min averages above the limit of detection for NO$_x$, NO$_y$, and PAN, ∼46% of the data show significant odd NO$_y$ levels (i.e. cannot be accounted for by total measurement uncertainty alone). What is particularly striking about the NO$_y$ levels is that odd NO$_y$ levels can be very large over winter (Fig. 3). The monthly mean odd NO$_y$ reached 110 ± 37 pmol mol$^{-1}$ (mean ± uncertainty) in February 2010. The peak in odd NO$_y$ was in March 2009 at 126 ± 48 pmol mol$^{-1}$; however, the data may be bias as there were only 34 coincident NO$_x$, NO$_y$, and PAN measurements in March 2009, whereas there were between ∼270 and 1200 coincident measurements during the other months. Odd NO$_y$ decreased to ∼20–60 pmol mol$^{-1}$ over the summer months; however, this still accounts for up to ∼40% of the total NO$_y$ during this period. Snowfall rates increase during the summer months at Summit (Dibb and Fahnestock, 2004); therefore, an increase in deposition of water-soluble species such as HNO$_3$ to the snowpack may result in the depletion of ambient odd NO$_y$ in the summer. The increase in solar radiation may also play an important role in the reduction of odd NO$_y$ species in the summer. Solberg et al. (1997) observed a decrease in odd NO$_y$ with increasing solar UV radiation in Spitsbergen, Norway. The authors suggested that species such as HONO, HNO$_3$, NO$_3$, N$_2$O$_5$, HO$_2$NO$_2$, and alkyl nitrates may contribute to NO$_y$ over the winter with the impact reducing in spring due to an increase in photolysis. A study on the seasonal variability of alkyl nitrates at Summit in 1998–1999 found that the light C$_1$–C$_4$ alkyl nitrates peaked over winter (Swanson et al., 2003). The monthly mean total C$_1$–C$_4$ mole fraction during February 1999 was 33 pmol mol$^{-1}$. Assuming a similar level during February 2010, alkyl nitrates would account for ∼1/3 of the odd NO$_y$ observed during this month. Taking into account measurement uncertainties, there remains, therefore, a large fraction of NO$_y$ unaccounted for over winter.

Results from this study show, despite the odd NO$_y$ levels decreasing from late spring to summer, odd NO$_y$ species can contribute over twice as much as NO$_x$ to the total NO$_y$ in summer. To investigate the source of the odd NO$_y$ species and the possible impact from snowpack photochemistry, we analyzed the ambient diurnal variability of NO$_x$, NO$_y$, and odd NO$_y$ at Summit separately for March, April, May, and June 2008–2010 (Fig. 4).

A clear diurnal cycle is observed in the NO$_x$, NO$_y$, and odd NO$_y$ measurements in April and May. The NO$_x$ diurnal cycle shows a minimum in the morning and peaks after solar noon. Diurnal cycles for NO and NO$_2$ are given separately in the Supplement and show NO similarly peaking around solar noon. In contrast, NO$_2$ levels reach a maximum overnight. The observed NO$_x$ cycle is in agreement with the cycle observed at Summit during summer 2000 (Honrath et al., 2002), which was attributed to photochemical reactions in the snowpack and enhanced vertical mixing during the day. Studies have shown that diurnally varying radiation at Summit can result in heating at the surface and the development of unstable to near-neutral conditions (Cullen and Steffen, 2001) and increasing boundary layer depths during the day (Helming et al., 2002; Cohen et al., 2007), with entrained air having the effect of diluting snowpack emissions near the surface. Consequently, increased mixing may offset increased daytime surface fluxes or production rates and dampen or offset diurnal concentration cycles in the surface layer.

The amplitude of the diurnal cycles (determined as the difference between the minimum and maximum 2 h median values) was always greater for NO$_x$ than NO$_y$ from March to June. The NO$_x$ and NO$_y$ amplitudes were the largest in May, with values of 35 and 22 pmol mol$^{-1}$, respectively. The diurnal cycle for odd NO$_y$ in May was 20 pmol mol$^{-1}$ peaking around solar noon (Fig. 4k), suggesting a photochemically
Figure 4. Average diurnal cycle of ambient NOy (a, b, c, d), NO3 (e, f, g, h), and odd NOy (i, j, k, l) measured at Summit for the months March (1st column), April (2nd column), May (3rd column), and June (4th column) 2008–2010. Median ambient levels observed each day were subtracted, to remove any impact from day to day variability. The median and mean of the data are represented by a horizontal line, and filled black circle, respectively; the box indicates the middle 67% of the data; and the vertical whiskers indicate the 5th and 95th percentile of all the data. Times are shown as local time (UTC). The amplitude of the diurnal cycle (given as the difference between the lowest and highest 2 h median values, in pmol mol−1) is noted on each subplot.

produced odd NOx species may be present. It has been hypothesized that HNO3 and HONO may account for some of the NOx diurnal variability at Summit (Ford et al., 2002). At Neumayer, Antarctica, the diurnal variability in NOy was attributed to both boundary layer changes and snowpack–air exchange of gases (Weller et al., 1999; Grannas et al., 2007). There is a possible contribution to the odd NOx diurnal cycle in the summer at Summit from long-range transport of pollution. Reactive nitrogen species such as HNO3 and alkyl nitrates as these species have previously been observed in anthropogenic and biomass burning plumes in the Arctic (Liang et al., 2011; Wespess et al., 2012) and the downward transport of pollution from aloft due to a growing boundary layer may result in a daytime maxima in NOx and NOy, which then decreases at night due to surface uptake.

Ambient HNO3 and HONO have been measured at Summit during a number of spring and summer campaigns. Levels of HNO3 are typically on the order of a few tens of pmol mol−1, and HONO levels are lower with mole fractions of ∼10 pmol mol−1 or less (Dibb et al., 1994, 1998, 2002, 2007; Honrath et al., 1999; Ford et al., 2002; Yang et al., 2002; Chen et al., 2007; Liao et al., 2011). Median mixing ratios of HNO3 and HONO measured during May and June 2010, with a mist chamber/ion chromatography (MC/IC) system (∼1.5 m above the snowpack), were 7 and 13 pmol mol−1, respectively (J. E. Dibb and M. G. Hasting, personal communication, 2014). Note that HONO measurements by MC/IC in polar regions should be viewed as an upper limit to the true value, due to potential interferences from other species (Chen et al., 2004; Liao et al., 2006). A direct comparison of HNO3 and HONO with calculated odd NOy levels is not possible for 2010 because PAN measurements were unavailable after April. Monthly mean odd NOy levels calculated for May and June in 2009 were 43 and 36 pmol mol−1, respectively. Summer levels of alkyl nitrates are expected to be low at Summit with Swanson et al. (2003) measuring monthly mean levels (total C1–C4) of ∼10–20 pmol mol−1 between May and August. Assuming alkyl nitrate levels of ∼15 pmol mol−1 during May/June, the sum of HNO3 + HONO + alkyl nitrates are comparable to the odd NOy levels measured at Summit, in May/June (when considering measurement uncertainties). Particulate nitrate may also contribute a small amount to the total NOy; however, this contribution is expected to be small as ambient p-NOy levels are typically lower than HNO3 at Summit (Dibb et al., 1994).

HO2NO2 may also contribute to the odd NOy measured at Summit. The partitioning of HO2NO2 between air and ice has a strong temperature dependence (Ulrich et al., 2012) and a recent study in Antarctica, during polar winter, has shown that absorbed HO2NO2 can be emitted into the atmosphere above the snowpack when temperatures increase (Jones et al.,...
2014). HO$_2$NO$_2$ was also investigated as a possible source of NO$_x$ during the Tropospheric Ozone Production about the Spring Equinox (TOPSE) campaign Stroud et al. (2003). Modelled HO$_2$NO$_2$ during February was comparable to the observed NO$_x$ deficit of 60 pmol mol$^{-1}$ between 3 and 6 km; however, the authors also note that the modelled HO$_2$NO$_2$ increased during May, whilst the observed NO$_x$ deficit had decreased. It is apparent that further measurements are required to determine both the species and sources of this odd NO$_x$ observed at Summit during winter.

In summary, the seasonal and diurnal cycles of NO$_x$ and NO$_y$ at Summit show that PAN is the dominant NO$_y$ species in this region year-round; however, NO$_x$ is an important contributor to the NO$_y$ budget in summer. Local thermal decomposition of PAN is unlikely at Summit, due to the cold temperatures; therefore, the NO$_x$ observed during summer is likely to be the result of upward emissions from the snowpack. Long-range transport may also play a role, as Summit is also influenced by pollution plumes that are mixed into the boundary layer (discussed further in Sect. 3.2). During spring and summer the NO$_x$ observed at Summit can be attributed primarily to PAN, NO$_y$, HONO, and HNO$_3$. Over winter, PAN is the dominant NO$_x$ species, and alkyl nitrates are also likely to contribute to the observed NO$_y$; however, there is still a large fraction of NO$_y$ that is unaccounted for over winter, and requires further investigation.

### 3.1.2 Non-methane hydrocarbons

Figure 5 shows the results for the C$_2$–C$_5$ alkane NMHC measured during 2008–2010 at Summit in ambient air. Measurements of primarily firm air conducted with this system were presented by Helmig et al. (2014). NMHC show a strong seasonal cycle with maximum mole fractions during the winter and early spring period and a rapid decline towards the summer, due to an increase in photochemical processing. The monthly averages for the C$_2$–C$_3$ NMHC are given in Table 2. During the summer period, measured mole fractions of the heavier NMHC were below or close to the detection limit. As expected, the phase of each NMHC is shifted due to the rate of reaction with OH. The lightest of the NMHC shown in Fig. 5a, ethane (C$_2$H$_6$), peaks in March with a monthly mean of 2100 ± 151 pmol mol$^{-1}$ (mean ± 1σ) in 2009, and 1835 ± 174 pmol mol$^{-1}$ in 2010, and declines to a minimum of ∼600 pmol mol$^{-1}$ in July/August. Heavier NMHC have lower mole fractions, peak earlier in the year, and reach a minimum earlier in summer due to their faster rate of reaction with OH.

A comparison between the two winters shows monthly mean NMHC levels were consistently higher from November 2008 to March 2009 when compared to November 2009 to March 2010. However, the standard deviation values during the second winter are typically higher, suggesting greater variability in the NMHC levels in 2009–2010. The non-averaged NMHC data (Fig. S6, Supplement), show there is considerable short-term variability in the NMHC mole fractions superimposed on the seasonal cycle, in particular during the winter months. Short-term elevated NMHC levels were observed during both winters and indicate fast transport of polluted air masses to the site.

The accumulation of O$_3$ precursors, such as nitrogen oxides and NMHC over winter has been suggested as a potential in situ source of O$_3$ that may contribute to the tropospheric O$_3$ peak (e.g. Penkett et al., 1993; Honrath et al., 1996; Monks, 2000; Blake et al., 2003). Results from studies during the TOPSE Atlas et al. (2003), have shown that the photochemical production of O$_3$ is important in the Arctic troposphere and can contribute more to the springtime O$_3$ budget than O$_3$ influx from stratosphere–troposphere exchange (Browell et al., 2003; Emmons et al., 2003).

The seasonality of NMHC can provide some insight into the potential for the photochemical production of O$_3$ in the Arctic troposphere during spring. Measurements of NMHC and O$_3$ during TOPSE show that within the mid-troposphere, total NMHC decreased by ∼6.2 ppbC from February to May, and that O$_3$ increased by ∼16 ppbv during the same
Table 2. Monthly statistics for NMHC measured at Summit from 2008 to 2010.

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<th>Propane (pmol mol⁻¹) Mean ± SD</th>
<th>i-Butane (pmol mol⁻¹) Mean ± SD</th>
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period (Blake et al., 2003). The data from this study show similar results for NMHC, with the sum of the C₂–C₆ NMHC decreasing by ~ 4.5 ppbv from February to May. The magnitude of the O₃ increase, at ~ 8 ppbv, is smaller than observed during TOPSE; however, the photochemical processing of NMHC in spring may contribute to the spring time peak of O₃ over Greenland.

3.2 Variability in ozone and its precursors from anthropogenic and biomass burning emissions

In Sect. 3.1, the seasonal cycles for O₃ precursors at Summit were discussed. Short-term enhancements in O₃ precursor levels indicated that the boundary layer over Summit was often impacted by polluted air masses from lower latitudes. In this section the interannual and short-term variability in the measured species at Summit, from 2008 to 2010, as a result of changes in transport pathways and the relative source contributions of pollutants from North America, Europe, and Asia are investigated.

Figure 6 shows the total monthly averaged FLEXPART tracer simulations for BC_{anthro} and BC_{fire} tracers for 2008, 2009, and 2010. The data show that anthropogenic pollution impacts started to increase in November, peaking over the dark winter period, and then decreased in late spring. The total BC_{anthro} tracer at Summit is dominated by emissions from North America, Europe, and Asia. Source contributions from each continent can vary month-to-month and year-to-year; however, European emissions were typically the largest contributor to the total monthly BC_{anthro} during the winter period, when the polar front recedes northwards. During winter (NDJFM) 2008–2009, 56 % of the total BC_{anthro} originated from Europe, with 33 % from North America, and 11 % from Asia. The following winter (NDJFM 2009–2010), European emission were higher at 69 %, with North America and Asia contributing 19 and 11 %, respectively. A very small contribution (<1 %) of the total anthropogenic tracer originated from other continents. Figure 6 shows that the contribution to the total BC_{anthro} from Asia is low year-round. Hirdman et al. (2010) investigated the contribution of pollutants from different sources to various Arctic surface sites from 2002 to 2007. The authors show that Summit is less sensitive to emissions from the surface in the Arctic region, than low elevation surface stations, and air masses transported to Summit from outside of Greenland are likely to originate from Europe and North America.
Figure 6. Bar chart showing the total monthly BC\textsubscript{anthro} (top) and BC\textsubscript{fire} (bottom) tracer from FLEXPART for 2008, 2009, and 2010. The different colours represent the contributions from North America, Europe, and Asia as shown in the legend.

In contrast to the anthropogenic tracer, North American emissions dominated the total BC\textsubscript{fire} tracer, with 69% of the total BC\textsubscript{fire} tracer originating from this region from July 2008 to July 2010. This result is expected as air masses arriving at Summit typically originate from North America during the summer months (Kahl et al., 1997). Previous studies have also shown that North America is a major source of biomass burning emissions transported to Summit (e.g. Whitlow et al., 1994; Legrand and de Angelis, 1996; Fuhrer and Legrand, 1997; Alexander et al., 2004; Stohl et al., 2006).

The high Asian contribution to the total BC\textsubscript{fire} tracer in spring 2008 is in agreement with observations during the PolarCAT campaigns in 2008, when a number of biomass burning plumes were observed from Siberia (Law et al., 2014, and references therein). During the measurement period, however, the contribution from Asian biomass burning emissions at Summit was low, compared to North America. European biomass burning emissions were also low, except during a period in March, when European fires dominated the total BC\textsubscript{fire} tracer. In 2008 and 2010 the contribution from total biomass burning emissions peaked in August. In 2009, however, there was no significant peak in biomass burning emissions at Summit, and BC\textsubscript{fire} tracer levels during the summer were much lower overall.

Variability in BC\textsubscript{fire} tracer can be a result of a change in emissions or transport pathways. Overpass and cloud-corrected MODIS Terra (MOD14CMH) and Aqua (MYD14CMH) fire pixel counts from the Climate Modelling Grid (CMG) product were downloaded from the University of Maryland ftp server (ftp://fuoco.geog.umd.edu) to investigate the variability in fire emissions. The monthly total fire pixel count and total fire radiative power (FRP), over the zonal region 40–75° N, were determined from March to September for 2008, 2009, and 2010 separately. The results (Fig. S7a–d, Supplement) show that there was a reduction in total fire counts and FRP in summer 2009, when compared to 2008 and 2010.

In the next two sections, the impact of anthropogenic and biomass burning emissions on O\textsubscript{3} and O\textsubscript{3} precursors are investigated.

### 3.2.1 Anthropogenic impacts

To investigate the source of the observed variability and the impact on ozone precursor levels, the anthropogenic tracer from FLEXPART retroplume simulations (BC\textsubscript{anthro}) and NMHC emissions ratios were used to determine changes in the transport pathways and relative source contributions of anthropogenic emissions from different continents.

An event with pollution transport was defined as identified when the BC\textsubscript{anthro} tracer was greater than the 75th percentile of the total BC\textsubscript{anthro} during the 2 year measurement period (corresponding to BC\textsubscript{anthro} > 0.0082 pmol mol\textsuperscript{-1}) for a minimum of 12 h. The FLEXPART temporal resolution for backward simulations is 3 h, so identifying events when the BC\textsubscript{anthro} was enhanced for at least 12 h ensured that significant polluted air masses impacted the site. Using these thresholds, 85 anthropogenic pollution events were observed during the measurement period. Details for each event, including start date, FLEXPART tracer levels, and trace gas enhancements are presented in Table S2 in the Supplement. The mean weighted age of the plume was also calculated for each FLEXPART retroplume, to determine the typical transport time for the events. As shown in Fig. 6, anthropogenic impacts are observed year-round; however, during the summer anthropogenic pollution events can be mixed with biomass burning emissions. Therefore, for this study, the focus is on events during November and March when FLEXPART anthropogenic tracer levels peaked and biomass burning emissions were typically low. In total, 52 events were identified during the periods November to March 2008–2009 (events 12–37) and November to March 2009–2010 (events 55–80). Typical mean transport times for polluted air masses in winter were ~11–12 days.

During each event, the mean ΔPAN, ΔNO\textsubscript{y}, ΔNO\textsubscript{x}, ΔO\textsubscript{3}, and ΔC\textsubscript{2}H\textsubscript{6} were calculated, where Δ is the enhancement above background levels (determined as the 20th percentile for each species, for each month and year). Care must be taken when defining a background level as the enhancement ratios are strongly dependent on this value. Initial analyses used a single background value for each month, determined from the 20th percentile of data from all years; however, this resulted in many negative or low enhancement ratios which were not consistent with the observations. A background value for each individual month and year was found to be more appropriate, as these levels change slightly from year to year. For the 52 anthropogenic events identified during winter, the median (and range) value of the enhancements were 38 (−22 to 295) pmol mol\textsuperscript{-1} for PAN, 6.6 (−1.9 to 88) pmol mol\textsuperscript{-1} for NO\textsubscript{y}, 67 (−17 to 334) pmol mol\textsuperscript{-1} for...
NO$_y$, 1.2 (−6.7 to 7.6) nmol mol$^{-1}$ for O$_3$, and 142 (−245 to 1119) pmol mol$^{-1}$ for C$_2$H$_6$.

Tests were performed to determine whether temporal mismatches occurred between the simulated plume arrivals and peaks in the measured species as a result of errors in the wind fields used in FLEXPART. Each event identified above was extended in time by 3, 6, 9, and 12 h prior to and after the event, and new $\Delta$ values for the O$_3$ precursors were calculated. The analyses showed that extending the event window did not have a large impact on the mean $\Delta$ values when mole fractions were low; i.e. levels were typically near the background already and extending the event time did not result in additional plumes being captured. During those events with large peaks in O$_3$ and precursors, $\Delta$ values decreased, with increasing window lengths as a result of capturing lower background levels. Thus, the event times calculated using the original threshold appear to be in agreement with the pollution plume arrival times.

Over the two winters, negative mean $\Delta$NO$_y$ values were observed during three events (event 72, 78, and 79), negative $\Delta$PAN during four events (18, 25, 72, and 73) and negative $\Delta$C$_2$H$_6$ during four events (64, 72, 73, and 79). Negative $\Delta$ values were associated with (1) events when the peak BC$_{anthro}$ level was low (relative to BC$_{anthro}$ peak levels during all events), indicating a small impact from anthropogenic pollution; (2) missing data points during the event; or (3) events when small enhancements in PAN and NO$_y$ occurred; however, low mole fractions were also observed during the same event period.

Analyses of $\Delta$O$_3$ and FLEXPART BC tracer masses show decreases in O$_3$ ($\Delta$O$_3$ < 0 nmol mol$^{-1}$), coinciding with anthropogenic pollution events, were observed during 20 periods in total from July 2008 to July 2010, with 16 of these events occurring between November and March, when sunlight is at a minimum. Negative $\Delta$O$_3$ do not always coincide with low O$_3$ precursor levels. For example, Fig. 7 shows a time series of measurements between 10 January and 2 March 2010. During this period, events with elevated ethane, NO$_y$, PAN, and NO$_x$ levels are clearly correlated with enhancements in the FLEXPART BC tracer, suggesting polluted air masses impacting the site; however, what is particularly striking is that O$_3$ mole fractions were low during some events (e.g. events 70, 71, and 74). Studies have shown that NMHC ratios can provide an indication of the photochemical aging of the air mass as the rate of reaction of different NMHC, and hence the ratio, is dependent on the amount of photochemical processing that occurs during transport (Parrish et al., 2004; Helmig et al., 2008; Honrath et al., 2008). High photochemical processing results in a decrease in the ln([propane] / [ethane]) ratio as propane reacts more readily with OH than ethane. The two low ozone events, during this period, with the largest ethane enhancements (70 and 71) on 22 and 29 January, coincided with enhancements in the ln([propane] / [ethane]) ratio, suggesting low photochemical processing, and fresher air masses; i.e. air that was subjected to more recent pollution before reaching at the site. It is difficult to obtain absolute air mass ages from NMHC ratios, due to dilution of the measured species during transport; however, the enhancements in the ln([propane] / [ethane]) ratio during the two pollution events coincide with sudden decreases in the mean weighted age of the plumes, estimated from FLEXPART. The vertically integrated emission sensitivity (also called the total column sensitivity, measured in nanosecond metres per kg) simulated by FLEXPART was used to determine the air mass transport pathway.

Figure 8a shows the total column sensitivity from FLEXPART on 23 January (during event 70). The air mass originated from northern Europe and resided in the lower ∼ 2 km during transport, until 1 day upwind when the air mass ascended over the surface of the Greenland ice sheet to the measurement site. Mean $\Delta$PAN and $\Delta$C$_2$H$_6$ levels during this event were 125 and 1036 pmol mol$^{-1}$, respectively, supporting the FLEXPART analyses indicating a polluted air mass was sampled (note there was no NO$_y$ data available during this period). Note that NO$_x$ was also enhanced (mean $\Delta$NO$_x$ = 87.7 pmol mol$^{-1}$) during this event. As the air mass resided in the lower atmosphere before transport to the site, where temperatures are warmer, the enhanced NO$_x$ observed is likely to be the result of PAN decomposition during transport. Ozone decreased, on average, 6.0 nmol mol$^{-1}$ below the monthly background (O$_3$(bkg) = 45 nmol mol$^{-1}$), and reached

![Figure 7](image-url)
a minimum level of 35.4 nmol mol\(^{-1}\). In contrast, a few hours later the transport patterns quickly changed, and the air masses sampled at Summit originated from high altitudes over North Canada, as shown in Fig. 8b. As a result, \(O_3\) levels increased and NO\(_x\), PAN, and ethane all decreased. Air originating from the high Arctic region was sampled at the site until 29 January, when the air transport moved southward, and air masses residing in the lower troposphere over North America transported polluted air to Summit (Fig. 8c). From 29 to 30 January (event 71), ethane, NO\(_y\), PAN, and NO\(_x\) all increased with mean enhancements of 1119, 184, 92.1, and 10.6 pmol mol\(^{-1}\), respectively. The mean \(\Delta O_3\) during event 71 was \(-0.7\) nmol mol\(^{-1}\), reaching a minimum \(O_3\) level of 39.6 nmol mol\(^{-1}\).

Background \(O_3\) levels at Summit are typically higher than those observed at lower elevation Arctic sites due to a stronger influence of transport from the stratosphere, a reduction in ozone depletion events from halogens, and low surface deposition rates (Helmig et al., 2007a, b; Hirdman et al., 2010). The decrease in \(O_3\) observed during winter/early spring is likely to be the result of titration of \(O_3\) by NO within the sampled air mass soon after emission (Eneroth et al., 2007; Hirdman et al., 2010) and reduced mixing with the background air. FLEXPART retroplume analyses indicate that low \(O_3\) events over winter typically coincide when sampling air masses originating from either Europe or North America, which have resided in the lower troposphere and/or are quickly transported over the Greenland ice sheet to the measurement site (examples of FLEXPART retroplumes are shown in Fig. S8a–e in the Supplement). In contrast, periods identified during winter as pollution events with positive \(O_3\) enhancement values often occurred when the air masses resided in the mid-troposphere during transport to the site (Fig. S9a–e), thus allowing for greater mixing with air from high tropospheric or stratospheric origin.

To investigate interannual variability in emissions and their impact on \(O_3\) and \(O_3\) precursors, a comparison of the enhancement values for \(\Delta O_3\), \(\Delta PAN\), and \(\Delta C_2H_6\) was made over the two winters (no comparison was made for NO\(_y\) or NO\(_x\) as measurements were limited during winter 2008–2009). Results show that enhancements in \(O_3\) and \(O_3\) precursors during events were typically higher dur-
ing the second winter season, with median (mean) values of 12 (37) pmol mol\(^{-1}\), 126 (162) pmol mol\(^{-1}\), and 0.7 (1.2) nmol mol\(^{-1}\) in 2008–2009, and 51 (56) pmol mol\(^{-1}\), 178 (237) pmol mol\(^{-1}\), and 1.4 (1.3) nmol mol\(^{-1}\) in 2009–2010, for \(\Delta\)PAN, \(\Delta C_2H_6\), and \(\Delta O_3\), respectively. FLEXPART analyses, presented in Sect. 3.2, show that European emissions contributed a greater amount to the anthropogenic air masses arriving at Summit in winter 2009–2010 compared to the previous year; thus, the higher enhancements are consistent with expectation that European emissions have a larger impact on the Arctic lower atmosphere in winter than North America or Asia.

In this study, the impact from anthropogenic emissions, as identified through FLEXPART retroplume analyses, were the primary focus. However, enhancements in the measured species were also observed during periods which are not correlated with pollution events simulated by FLEXPART. For example, on 15 February and 24–28 February 2010, (Fig. 7), enhancements in PAN, NO\(_y\), and ethane are observed that do not coincide with high FLEXPART BC\(_{anthro}\) tracer levels. The BC\(_{fire}\) tracer from FLEXPART during this period is low, suggesting that the event was not the result of biomass burning emissions. The retroplume shows the air masses from these two events were transported over northern Canada and remained in the Arctic for many days before arriving at Summit (not shown here). It is unlikely that the air sampled was from stratospheric origin, as ethane levels were high and ozone decreased during these events. It is possible that there may be missing sources in FLEXPART, or that the pollution originated prior to the 20 day simulation.

Enhancements in O\(_3\) precursors that do not coincide with high FLEXPART tracer simulations were also observed during the spring months; however, O\(_3\) was typically enhanced during these events, suggesting a different source than those in February 2010. FLEXPART analyses indicate the air masses were well aged, therefore, mixing with air masses originating from the upper-troposphere and/or stratosphere is possible. For example, the highest PAN and NO\(_y\) levels during spring 2010 were observed between 22 and 24 April, when FLEXPART retroplumes show that the air masses were transported between 3300 and 4300 m a.s.l. (Fig. 9). During this same period, ethane and O\(_3\) were also high. This event may be an example of tropospheric air, with high levels of NMHC, mixing with stratospheric air rich in O\(_3\) and NO\(_y\), resulting in PAN formation (Liang et al., 2011). The determination of the sources that result in enhancements in O\(_3\) and its precursors, that cannot be attributed to pollution transport, requires further investigation.

The study here has focused on anthropogenic impacts during the winter period as this is when FLEXPART anthropogenic tracers peak; however, it should be noted that high \(\Delta O_3\) events occurred in the summer months, in particular between May and August. Some of the anthropogenic air masses were mixed with BB plumes (as discussed in the next section); however, the largest enhancement in O\(_3\), during the measurement period (\(\Delta O_3 = 15.0 \text{ nmol mol}^{-1}\)), occurred from 4 to 8 June 2009, as a result of transport of anthropogenic emissions from Europe to Summit in \(\sim 10\) days. O\(_3\) precursors were also enhanced during this event, indicating that further O\(_3\) production may be possible during subsequent transport.

### 3.2.2 Biomass burning impacts

The extended whiskers shown on the plots in Fig. 1 indicate a large amount of variability in the O\(_3\) precursors during the summer months. Radiation, surface emissions, boundary layer height, and changes in air mass sampling may all contribute to the variability observed. Anthropogenic emission impacts tend to be lower in the summer as a result of reduced transport from source regions, as shown in Fig. 6; however,
pollution from anthropogenic, and especially, BB emissions can still impact the centre of Greenland (Stohl, 2006), resulting in elevated mole fractions for short periods (Stohl et al., 2006). Studies based on aircraft measurements and models during the ARCTAS campaigns in both spring and summer 2008 show that NO\textsubscript{x}, NO\textsubscript{y}, PAN, and hydrocarbons levels can be elevated in biomass and anthropogenic plumes transported to the Arctic (e.g. Alvarado et al., 2010; Singh et al., 2010; Hornbrook et al., 2011; Liang et al., 2011).

The BC\textsubscript{fire} tracer from FLEXPART was used to identify periods at Summit that were potentially impacted by BB emissions. Potential inaccuracies with the FLEXPART simulation of transport pathways, fire identification, and tracer emission uncertainties may result in BB events being under or overestimated; however, FLEXPART has been successfully used to identify long-range transport of biomass burning emissions in many studies (e.g. Brioude et al., 2007; Stohl et al., 2007; Lapina et al., 2008; Quennehen et al., 2011, 2012; Schmale et al., 2011; Cristofanelli et al., 2013).

Biomass burning events were characterized as having a FLEXPART BC\textsubscript{fire} tracer >90th percentile (≈7 pmol mol\textsuperscript{-1}). In total, 13 events were observed between July 2008 and July 2010 ranging in duration between 12 and 252 h. Details regarding the start date, duration, mean plume age, FLEXPART tracer levels and trace gas levels for each event are shown in Table 3. A more conservative threshold was applied here than for the anthropogenic emissions in Sect. 3.2.1; consequently, the events identified had significant BB impacts but small anthropogenic signatures. Of these 13 events, two were identified as having potentially high anthropogenic signatures (BC\textsubscript{anthro} >75th percentile) and were most likely plumes of mixed anthropogenic and biomass burning emissions, six events were identified as having medium anthropogenic signatures (75th percentile > BC\textsubscript{anthro} >50th percentile) and the remaining five events were classified as having low anthropogenic signatures. The source contribution from FLEXPART shows that the majority of the BB events (9 out of 13) originated in North America, with the BB events originating in Europe all occurring in March 2009.

Table 3. Biomass burning events and mean enhancements in trace gases measured at Summit.

<table>
<thead>
<tr>
<th>Event</th>
<th>Start date</th>
<th>Event length (h)</th>
<th>BC\textsubscript{fire} (pmol mol\textsuperscript{-1})</th>
<th>ΔO\textsubscript{3}\textsuperscript{a} (nmol mol\textsuperscript{-1})</th>
<th>ΔPAN (pmol mol\textsuperscript{-1})</th>
<th>ΔNO\textsubscript{x} (pmol mol\textsuperscript{-1})</th>
<th>ΔNO\textsubscript{y} (pmol mol\textsuperscript{-1})</th>
<th>ΔC\textsubscript{2}H\textsubscript{6} (pmol mol\textsuperscript{-1})</th>
<th>Plume age\textsuperscript{b} (days)</th>
<th>Source\textsuperscript{c}</th>
<th>BC\textsubscript{anthro}</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>25 Jul 2008</td>
<td>33</td>
<td>30.6</td>
<td>-4.6</td>
<td>6.3</td>
<td>5.7</td>
<td>12.6</td>
<td>48.6</td>
<td>9 NA</td>
<td>med</td>
<td>NA</td>
</tr>
<tr>
<td>2</td>
<td>1 Aug 2008</td>
<td>15</td>
<td>25.0</td>
<td>7.0</td>
<td>83.5</td>
<td>13.2</td>
<td>141</td>
<td>-</td>
<td>14 NA</td>
<td>med</td>
<td>NA</td>
</tr>
<tr>
<td>3</td>
<td>3 Aug 2008</td>
<td>252</td>
<td>90.7</td>
<td>10.5</td>
<td>8.5</td>
<td>21</td>
<td>-</td>
<td>-107</td>
<td>12 EU</td>
<td>med</td>
<td>NA</td>
</tr>
<tr>
<td>4</td>
<td>15 Mar 2009</td>
<td>60</td>
<td>58.6</td>
<td>-0.5</td>
<td>2.1</td>
<td>-</td>
<td>-</td>
<td>-11.3</td>
<td>15 EU</td>
<td>med</td>
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</tr>
<tr>
<td>5</td>
<td>18 Mar 2009</td>
<td>33</td>
<td>23.1</td>
<td>3.8</td>
<td>-1.1</td>
<td>-</td>
<td>-</td>
<td>-11.1</td>
<td>16 EU</td>
<td>low</td>
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</tr>
<tr>
<td>6</td>
<td>21 Mar 2009</td>
<td>21</td>
<td>17.6</td>
<td>3.8</td>
<td>-20.9</td>
<td>-</td>
<td>-</td>
<td>11.6</td>
<td>16 EU</td>
<td>low</td>
<td>NA</td>
</tr>
<tr>
<td>7</td>
<td>27 May 2009</td>
<td>30</td>
<td>17.7</td>
<td>10.7</td>
<td>71.6</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>16 AS</td>
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</tr>
<tr>
<td>8</td>
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<td>19.9</td>
<td>4.6</td>
<td>67.2</td>
<td>6.2</td>
<td>108</td>
<td>-9.6</td>
<td>13 NA</td>
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<td>NA</td>
</tr>
<tr>
<td>9</td>
<td>18 Jul 2009</td>
<td>15</td>
<td>13.5</td>
<td>13.0</td>
<td>58.2</td>
<td>22.6</td>
<td>135</td>
<td>25.5</td>
<td>14 NA</td>
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<td>NA</td>
</tr>
<tr>
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<td>16 Aug 2009</td>
<td>18</td>
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<td>-4.1</td>
<td>3.3</td>
<td>16.0</td>
<td>11.8</td>
<td>-</td>
<td>18 NA</td>
<td>low</td>
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<tr>
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<td>12</td>
<td>9.5</td>
<td>5.7</td>
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<td>7.5</td>
<td>65.4</td>
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<td>17 NA</td>
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<td>NA</td>
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<tr>
<td>12</td>
<td>7 Jun 2010</td>
<td>27</td>
<td>11.3</td>
<td>8.9</td>
<td>-</td>
<td>5.6</td>
<td>105</td>
<td>13.1</td>
<td>13 NA</td>
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<td>NA</td>
</tr>
<tr>
<td>13</td>
<td>18 Jul 2010</td>
<td>51</td>
<td>27.4</td>
<td>4.3</td>
<td>-</td>
<td>4.7</td>
<td>97.3</td>
<td>27.8</td>
<td>9 NA</td>
<td>high</td>
<td></td>
</tr>
</tbody>
</table>

\textsuperscript{a} A represents the enhancement over the background level (background is 20th percentile of each species for each month and year). \textsuperscript{b} Mean weighted age from FLEXPART. \textsuperscript{c} Primary BB source. NA is North America, EU in Europe, AS is Asia. \textsuperscript{d} Indicates potential contribution from anthropogenic pollution. Low: BC\textsubscript{anthro} < 50th percentile; Med: BC\textsubscript{anthro} < 75th percentile; high: BC\textsubscript{anthro} > 75th percentile.
therefore, enhancements observed in the gas species may also be the result of anthropogenic pollution.

Enhanced PAN, NOx, and NOy were observed during two BB events on July 17 and 18 July 2009 (event eight and nine). Plumes during both events were transported at lower altitudes (~1950–2900 m) than the BB plume in August 2008, and had low anthropogenic signatures. Mean ΔO3 levels were 4.6 and 13.0 nmol mol−1 during event numbers 8 and 9, respectively, suggesting the BB emissions resulted in photochemical production of O3 during transport to the site. Care must be taken when interpreting these results. All the BB plumes sampled during this study have long transport times from source region to the measurement site, with mean weighted plume ages for the events ranging between 9 and 18 days (median 14 days). These aged plumes will be well mixed with background air; therefore, separating the pollution impacts from background levels is challenging.

Aged plumes transported at higher altitudes have a higher probability of mixing with high ozone from stratospheric origin, which may contribute to the elevated O3 levels that were observed. For example, during event 7, FLEXPART indicates that an aged plume (mean age ∼16 days) originating from Asia, was transported over the North Pole between ∼3000 and 4200 m a.s.l. A similar transport pattern was observed during the POLARCAT campaign, during which a plume, originating from Asia, was lifted within a warm conveyor belt over the North Pole towards Greenland (Roiger et al., 2011). The plume was then transported into the lower stratosphere, resulting in a well mixed tropospheric–stratospheric air mass with high O3 levels. The plume during event seven was transported at lower altitudes; however, mixing with air from stratospheric origin is not impossible and may be the cause of the O3 (ΔO3 = 10.7 nmol mol−1) and PAN (ΔPAN = 71.6 pmol mol−1) levels observed.

BB plumes that are well mixed with background air can result in low precursor levels. In 2008 FLEXPART indicated that a BB event impacted the measurement site from 25 to 26 July (event 1). The total column sensitivity from the FLEXPART retroplume (see Fig. 11) shows that the air masses arriving at the site during this event originated from a region with BB sources over Canada and Alaska and were transported in the lower troposphere over the Arctic Ocean for ∼4 days before ascending to Summit. The mean enhancement for O3 during this event was −4.6 nmol mol−1 and O3 precursors have small enhancements. Similar results were observed in March 2009 (events 4, 5, 6), when ΔPAN values were low or negative and ΔO3 levels were below 4 nmol mol−1. During all three of these events, the air masses spent many days in the lower troposphere, over the Arctic ocean and the edge of the Greenland ice sheet, before ascending to Summit in 1–2 days. It is likely, during these events, the aged plumes were well mixed with marine air with low O3 and precursor mixing ratios.

4 Summary

These analyses of NOx, NOy, PAN, NMHC, and O3 from the high altitude GEOSummit station in Greenland show that PAN is the dominant species of NOy at the site year-round, with monthly mean contributions ranging from a minimum of 42% in the summer months to 76% in spring. However, the NO3 seasonal cycle does not follow that of PAN, due to significant contributions from NOx in the summer, and odd NO3 species during both summer and winter. We hypothesize that HNO3, HONO, and alkyl nitrates are the largest sources of odd NOy in the summer months. During the winter, mean monthly odd NO3 levels reach over 100 pmol mol−1. Approximately one-third of the odd NO3 is likely to be alkyl nitrates transported to the site. Previous studies have suggested that HONO, HNO3, NO3, N2O5, and HO2NO2 may also contribute to the wintertime NO3 levels.

Rapid changes in the origin of sampled air masses, from regions in Europe, North America, and the high lat-
FLEXPART tracer simulations indicated that biomass burning emissions transported to Summit during the summer in 2008–2010 primarily originated from North America. Biomass burning plumes from Europe were only present during a short period in March 2009. During 2009 measurements were performed over a full summer period; however, BB impacts were low in summer 2009, compared to 2008 and 2010. In total, 13 plumes were identified, and only five with low anthropogenic signatures. During the events, O$_3$ and precursor levels were typically enhanced within the BB plumes with $\Delta$O$_3$ levels up to 13.0 nmol mol$^{-1}$ and $\Delta$PAN, $\Delta$NO$_y$, and $\Delta$C$_2$H$_6$ levels enhanced by up to 83.5, 135 pmol mol$^{-1}$, and 113 pmol mol$^{-1}$, respectively. The results presented here show that BB plumes may potentially impact the O$_3$ at Summit and the Arctic region in general; however, due to the long transport times and limited number of BB air masses transported to Summit during the measurement period, a quantitative analysis of the impact of BB plumes on O$_3$ and O$_3$ precursor levels is difficult. Future studies, with coincident CO, O$_3$, and O$_3$ precursor measurements at Summit, would be valuable to determine enhancement ratios with respect to CO and evaluate the potential for ozone production in the region.

The data and results presented here show that the boundary layer at Summit is strongly impacted by both snowpack emissions and mixing of pollution plumes transported from North America and Europe. Anthropogenic emissions can impact the site year-round resulting in enhanced levels of NO$_y$ in the boundary layer. Knowledge of pollution impacts over the Greenland ice sheet is important for modelling studies, in particular those that are focusing on snowpack–air exchange of gas species. Enhanced NO$_y$ levels observed above the background during the events discussed here may have an important impact on snow photochemistry and the subsequent release of NO$_x$ due to the uptake of NO$_x$ species to the snowpack (Grammas et al., 2007, and references therein). Additionally, due to the stability of the Arctic free troposphere, the region is an effective reservoir for O$_3$ precursors. The high O$_3$ precursor mole fractions above background levels in spring and summer at Summit may have important implications for NO$_x$ and O$_3$ in the mid-latitudes during southerly flow of air masses (Hamlin and Honrath, 2002).

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